

Exploratory Combustion Synthesis: Amorphous Indium Yttrium Oxide for Thin-Film Transistors

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Supporting Information

ABSTRACT: We report the implementation of amorphous indium yttrium oxide (a-IYO) as a thin-film transistor (TFT) semiconductor. Amorphous and polycrystalline IYO films were grown via a low-temperature solution process utilizing exothermic "combustion" precursors. Precursor transformation and the IYO films were analyzed by differential thermal analysis, thermogravimetric analysis, X-ray diffraction, atomic force microscopy, X-ray photoelectron spectroscopy, and optical transmission, which reveal efficient conversion to the metal oxide lattice and smooth, transparent films. a-IYO TFTs fabricated with a hybrid nanodielectric exhibit electron mobilities of 7.3 cm² V⁻¹ s⁻¹ ($T_{\text{anneal}} = 300 \,^{\circ}$ C) and 5.0 cm² V⁻¹ s⁻¹ ($T_{\text{anneal}} = 250 \,^{\circ}$ C) for 2 V operation.

The past decade has seen major advances in the field of materials for unconventional electronics.¹ Products such as smart windows, large-area active-matrix organic lightemitting diode (AMOLED) displays, printable radiofrequency identification (RF-ID) tags, and flexible displays all require the understanding-based implementation of new electronic materials.² To this end, crystalline and amorphous metal oxide semiconductors have been of great interest because of the possibility of coupling optical transparency and mechanical flexibility with excellent electrical performance. Indeed, the thin-film transistor (TFT) performance of many oxides exceeds that of amorphous silicon (a-Si:H), and their stability rivals or exceeds that of typical organic semiconductors.³ While metal oxide films have conventionally been grown by vapor-phase techniques such as atomic layer deposition, pulsed laser deposition,^{3b} chemical vapor deposition, and RF sputtering,⁴ solution-phase processes such as spin-coating,⁵ dip-coating,⁶ chemical bath deposition,⁷ inkjet printing,⁸ etc. offer the attraction of low-cost, high-throughput, roll-to-roll processing. Thus, excellent films of $ZnO_{,9}^{,9}$ In₂ $O_{,9}^{,10}$ amorphous indium zinc oxide (a-IZO),¹¹ amorphous zinc indium tin oxide (a-ZITO),¹² and amorphous indium gallium zinc oxide (a-IGZO)¹³ can be grown from solution; however, the high required annealing/ densification temperatures are typically incompatible with flexible polymer substrates.¹⁴

We recently reported a low-temperature route to mixed metal oxide films using a "combustion" process in which the heat required for oxide lattice formation is provided by the large internal energies of the precursors (Figure 1A,B).¹⁵ In this method, an oxidizer and a fuel are combined in the precursor



Figure 1. (A) Oxidizer + fuel pair scheme for the combustion synthesis of IYO. (B) Qualitative energetics of a combustion synthesis. (C) Thermal analysis of $In_{1.95}Y_{0.05}O_3$ film formation from a combustion precursor.

mixture, and mild annealing suffices to initiate an exothermic reaction within the film, affording smooth, flexible, and electrically active films. While In_2O_3 , a-IZO, and amorphous zinc tin oxide (a-ZTO) films were demonstrated, the intriguing question arises as to whether this technique could be used for the exploratory synthesis of new oxide films, especially amorphous ones that might benefit from the low processing temperatures.

To achieve acceptable TFT performance (i.e., high mobilities and low carrier concentrations), small oxygen-binding cations such as Zn^{2+} or Ga^{3+} are typically introduced into In_2O_3 or SnO_2 hosts. Furthermore, mixing disparate metal ions promotes amorphous character and hence electrical uniformity, smooth surfaces, and flexibility due to the lack of grain boundaries. Recently, more strongly oxygen-binding group 3 and 4 cations, including Zr,¹⁶ Ti,¹⁷ Hf,¹⁸ and Sc,¹⁹ have been studied as matrix dopants to stabilize oxide lattices and suppress crystallization. Here we demonstrate a low-temperature route

Received:
 April 14, 2012

 Published:
 May 24, 2012

to amorphous indium yttrium oxide (a-IYO) films having minimal Y doping, using combustion processing. A ternary system such as IYO is preferred to a quaternary system to reduce the number of potential electron scattering sites in the lattice. Also, incorporation of strongly oxygen-binding Y³⁺ can further reduce the number of charge scattering centers through amorphization at minimal doping levels. To date, Y doping of In2O3 has been reported only in bulk systems in hightemperature catalyst²⁰ or ferroelectric²¹ syntheses, and recent theoretical work suggests that Y incorporation into In2O3 should widen the band gap for greater transparency while preserving electron mobility.²² We report here that as little as 2.5 mol % Y doping effectively suppresses In_2O_3 crystallization at 250 °C, affording TFTs with high mobilities (5.0 $\mbox{cm}^2\mbox{ V}^{-1}$ s^{-1}) and low operating voltages (2 V) on a self-assembled nanodielectric. Furthermore, IYO provides optical transparency superior to that of In2O3. Thin films of this novel semiconductor are very smooth when deposited by scalable spincoating.

In and Y combustion precursor solutions were prepared from the nitrate salts at a total metal concentration of 0.05 M with acetylacetone + ammonia as the fuel. The Y molar concentration was varied from 2.5 to 25% in solutions aged for 13 h. Thin IYO films (~12 nm thick) were deposited by spin-coating the combined precursor solutions onto Si(100)substrates, annealing at 250 or 300 °C in air for 20 min, cooling to 25 °C, and then repeating this process three times. The thermal analysis data in Figure 1C reveal that ignition of the combustion precursors occurs at ~200 °C for the entire composition range. X-ray diffraction (XRD) data for the resulting IYO films annealed at 250 °C (Figure 2A) show that even for small Y concentrations, all of the films are amorphous. Thus, Y efficiently disrupts the In₂O₃ lattice, promoting the formation of amorphous materials. For 300 °C film annealing (Figure 2C), Y doping at $\leq 5 \mod \%$ yields polycrystalline films



Figure 2. (A, C) XRD patterns for IYO films with varying Y content annealed at (A) 250 and (C) 300 °C. (B, D) AFM images of typical IYO films containing (B) 2.5 mol % Y at $T_{\text{anneal}} = 250$ °C and (D) 10 mol % Y at $T_{\text{anneal}} = 300$ °C.

with diffraction patterns matching that of In₂O₃, while Y doping at ≥ 10 mol % significantly disrupts the In₂O₃ lattice, yielding amorphous films. Similar trends were observed in the XRD of IYO films annealed at elevated temperatures [Figure S1 in the Supporting Information (SI)]. Surface characterization by atomic force microscopy (AFM) reveals very smooth, featureless films with root-mean-square (rms) roughnesses in the 0.19–0.28 nm range (Figure 2B,D and Figures S2 and S3).

X-ray photoelectron spectroscopy (XPS) was used to analyze IYO surface impurities, carbon contamination, and oxygen binding (Figures S4 and S5). A survey scan reveals no organic impurities, in accord with previous combustion studies,¹⁵ and shows slow evolution of the Y 3p and 3d features at \sim 300 and 155 eV, respectively, as the amount of Y is increased. Detailed analysis of the carbon 1s spectrum reveals a slight increase in the peak centered at \sim 288.7 eV as the Y content is increased. This likely reflects the substantial basicity of Y2O3, inducing reactions with CO2 or other carbonaceous compounds in ambient air.²³ Evaluation of the O 1s features reveals that as the Y content is increased, a corresponding decrease in the 530.7 eV feature occurs, which is ascribable to the metal-oxygenmetal lattice.²⁴ Concurrently, surface or bulk M-OH species at 532.2 eV increase with increasing Y, in good agreement with the XRD analysis. All of the reported films exhibit excellent optical transparency (>90%) when deposited on glass (Figure S6).

TFTs were next fabricated on doped Si wafers with a thermally grown SiO₂ dielectric and Al source/drain electrodes deposited via thermal evaporation (channel length = 100 μ m and width = 5 mm) to afford the final bottom-gate/top-contact structures n⁺⁺Si/SiO₂(300 nm)/IYO/Al. Experimental details and *I*-*V* plots are presented in the SI. Figure 3A,B shows typical transfer and output plots, and Figure 4A,B shows TFT performance metrics for $T_{\text{anneal}} = 250$ and 300 °C, respectively, for varied Y doping concentrations. At lower growth temperatures, a maximum mobility of 0.75 cm² V⁻¹ s⁻¹ is achieved with a large current on/off ratio ($I_{\text{on}}/I_{\text{off}}$) of 10⁵-10⁶ and a threshold voltage (V_{T}) of 10 V. Notably, as the Y doping is increased, the mobility decreases significantly, and the devices become inactive at ≥ 20 mol % Y, which is not surprising since Y₂O₃ is a known high- κ dielectric.²⁶

Transistors annealed at 300 °C exhibit mobilities as high as ~5 cm² V⁻¹ s⁻¹ at \leq 5 mol % Y. However, when the Y doping is increased to 10 mol %, the films became amorphous by XRD yet still maintain an excellent mobility of 2.2 cm² V⁻¹ s⁻¹. This value is roughly twice those of a-Si:H and conventional solution-processed a-IGZO films annealed at 400 °C.²⁴ For both annealing temperatures, as the Y content is increased, the oxygen vacancy and free carrier concentrations decrease. As a result, I_{on}/I_{off} initially increases because of the reduced I_{off} while a similar I_{on} is maintained. Further increases in Y content depress $I_{\rm on}/I_{\rm off}$ as a result of the significant decrease in $I_{\rm on}$ from the mobility degradation. On the other hand, $V_{\rm T}$ is related to the free carrier concentration and trap density. At low Y content, $V_{\rm T}$ follows the trend expected from the decrease in carrier concentration, further confirming the role of oxygen vacancy suppression. However, as the Y content is increased, we observe an unstable trend in $V_{\rm T}$. We speculate that this result is related to reaction with the atmosphere, considering the basic nature of Y2O3.²³ Investigations of device passivation and interface reactions are underway.

Next, a-IYO TFTs were fabricated on high-*κ* dielectric zirconia-based self-assembled nanodielectric (Zr-SAND) films



Figure 3. Typical (A, C) transfer and (B, D) output plots for a-IYO films with 2.5 mol % Y and $T_{\text{anneal}} = 250 \text{ °C}$ on (A, B) a SiO₂ gate dielectric and (C, D) a Zr-SAND gate dielectric. (E) Schematics of the TFT and Zr-SAND structures.²⁵

(Figure 3E)²⁵ having a thickness of ~12 nm and a capacitance of 450 nF/cm² to demonstrate the compatibility of a-IYO with various dielectrics and low-voltage operation. Devices with the structure n⁺⁺Si/Zr-SAND/a-IYO/Al were fabricated at 250 °C/ 2.5 mol % Y and at 300 °C/10 mol % Y. Typical transfer and output plots (Figure 3C,D and Figure S9) exhibit saturation at low operating voltages and good pinch-off, with electron mobilities of 5.0 and 7.3 cm² V⁻¹ s⁻¹ for 250 and 300 °C, respectively, at operating voltages of only 2.0 V (Table S1 in the SI). The superior performance of Zr-SAND/a-IYO TFTs versus SiO₂/a-IYO TFTs is the result of the increased dielectric capacitance and decreased semiconductor–dielectric interface trap-state density.²⁷

These results demonstrate the first use of Y doping of In_2O_3 to generate amorphous semiconducting IYO films at low temperatures by combustion processing. The a-IYO TFTs exhibit excellent device performance when fabricated with Zr-SAND. This integration of combustion-derived amorphous oxide films with hybrid nanodielectrics points to future opportunities in novel materials design.



Figure 4. TFT data for IYO films having varied Y mol % annealed at (A) 250 and (B) 300 °C.

ASSOCIATED CONTENT

S Supporting Information

Experimental details, XRD of IYO at high T_{anneal} , XPS spectra, additional AFM and rms roughness data, representative transfer and output plots for all of the fabricated TFTs, and performance metrics for Zr-SAND/a-IYO TFTs. This material is available free of charge via the Internet at http://pubs.acs.org.

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Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

We thank ONR (MURI N00014-11-1-0690), AFOSR (FA9550-08-1-0331), the Northwestern Materials Research Science and Engineering Center (NSF DMR-1121262), and Polyera Corp. for support of this research. Microscopy studies were performed in the NIFTI and KECK II facilities of the NUANCE Center at Northwestern University, which is supported by NSF-NSEC, NSF-MRSEC, the Keck Foundation, the State of Illinois, and Northwestern University.

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